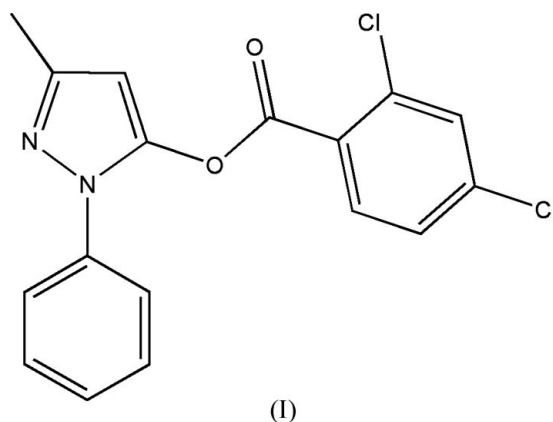


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janslee1103@yahoo.com.cn**Key indicators**Single-crystal X-ray study
 $T = 294$ K
Mean $\sigma(\text{C}-\text{C}) = 0.005$ Å
 R factor = 0.058
 wR factor = 0.178
Data-to-parameter ratio = 16.0For details of how these key indicators were automatically derived from the article, see <http://journals.iucr.org/e>.**3-Methyl-1-phenyl-1H-pyrazol-5-yl 2,4-dichlorobenzoate**In the title compound, $\text{C}_{17}\text{H}_{12}\text{Cl}_2\text{N}_2\text{O}_2$, the dihedral angle formed between the dichlorobenzene and pyrazolyl rings is $38.3(3)^\circ$, that between the phenyl and pyrazolyl rings is $46.8(3)^\circ$, and that between the two benzene rings is $69.4(3)^\circ$.Received 12 October 2005
Accepted 28 November 2005
Online 7 December 2005**Comment**The title compound, (I), and some of its previously reported analogs possess herbicidal and growth-regulating activities (Vasilev *et al.*, 1981) and anti-inflammatory properties (Terebenina *et al.*, 1980).In the molecular structure of (I) (Fig. 1), the dihedral angle formed between the dichlorobenzene and pyrazolyl rings is $38.3(3)^\circ$, that between the phenyl and pyrazolyl rings is $46.8(3)^\circ$, and that between the two benzene rings is $69.4(3)^\circ$.**Experimental**To a suspension of 1-phenyl-3-methyl-1H-pyrazol-5-one (Liu & Li, 2004; 0.36 g, 2 mmol), anhydrous sodium carbonate (0.11 g, 1 mmol), and a catalytic amount of tetrabutylammonium bromide in benzene (20 ml) and water (2 ml) was added dropwise 2,4-dichlorobenzoyl chloride (0.44 g, 2.1 mmol) in benzene (5 ml) over a period of half an hour at 283 K, and the reaction mixture was stirred at ambient temperature for an additional 1 h. The reaction was quenched by aqueous saturated sodium carbonate (10 ml), and the benzene layer was collected and evaporated under reduced pressure to yield 0.65 g of (I) as a colorless solid in 94% yield (m.p. 364.0–365.5 K). ¹H NMR (CDCl_3 , 500 MHz, p.p.m.): δ 7.85 (*d*, 1H, $J = 8.5$ Hz), 7.56–7.53 (*m*, 3H), 7.43 (*t*, 2H, $J = 7.5$ Hz), 7.34–7.31 (*m*, 2H), 6.28 (*s*, 1H), 2.36 (*s*, 3H); ¹³C NMR (CDCl_3 , 500 MHz, p.p.m.): δ 159.5, 149.1, 144.0, 140.0, 137.9, 136.3, 133.0, 131.7, 129.1 (2C), 127.5, 127.4, 125.6, 123.5 (2C), 95.9, 14.5. Suitable crystals were grown by evaporation from a solution in ethyl acetate and *n*-hexane (2:1 *v/v*).

Crystal data

C₁₇H₁₂Cl₂N₂O₂
M_r = 347.19
 Monoclinic, *P*2₁/*n*
a = 10.799 (3) Å
b = 10.385 (3) Å
c = 14.656 (4) Å
 β = 91.292 (4)°
V = 1643.3 (7) Å³
Z = 4

D_x = 1.403 Mg m⁻³
 Mo *K*α radiation
 Cell parameters from 2156 reflections
 θ = 2.4–21.7°
 μ = 0.41 mm⁻¹
T = 294 (2) K
 Prism, colorless
 0.24 × 0.20 × 0.18 mm

Data collection

Bruker SMART CCD area-detector diffractometer
 φ and ω scans
 Absorption correction: multi-scan (SADABS; Sheldrick, 1996)
T_{min} = 0.892, *T_{max}* = 0.930
 9043 measured reflections

3354 independent reflections
 1837 reflections with *I* > 2σ(*I*)
R_{int} = 0.044
 θ_{max} = 26.4°
h = -13 → 13
k = -6 → 12
l = -17 → 18

Refinement

Refinement on *F*²
R [*F*² > 2σ(*F*²)] = 0.058
wR (*F*²) = 0.178
S = 1.02
 3354 reflections
 209 parameters
 H-atom parameters constrained

$w = 1/[\sigma^2(F_o^2) + (0.0805P)^2 + 0.7399P]$
 where $P = (F_o^2 + 2F_c^2)/3$
 $(\Delta/\sigma)_{\text{max}} = 0.002$
 $\Delta\rho_{\text{max}} = 0.66 \text{ e \AA}^{-3}$
 $\Delta\rho_{\text{min}} = -0.48 \text{ e \AA}^{-3}$

All H atoms were included in the riding-model approximation, with C–H distances of 0.93 (aromatic) and 0.96 Å (methyl), and with *U_{iso}*(H) = 1.2*U_{eq}*(C_{arom}) and 1.5*U_{eq}*(C_{methyl}).

Data collection: SMART (Bruker, 1997); cell refinement: SAINT (Bruker, 1997); data reduction: SAINT; program(s) used to solve structure: SHELXS97 (Sheldrick, 1997); program(s) used to refine structure: SHELXL97 (Sheldrick, 1997); molecular graphics: SHELXTL (Bruker, 1997); software used to prepare material for publication: SHELXTL.

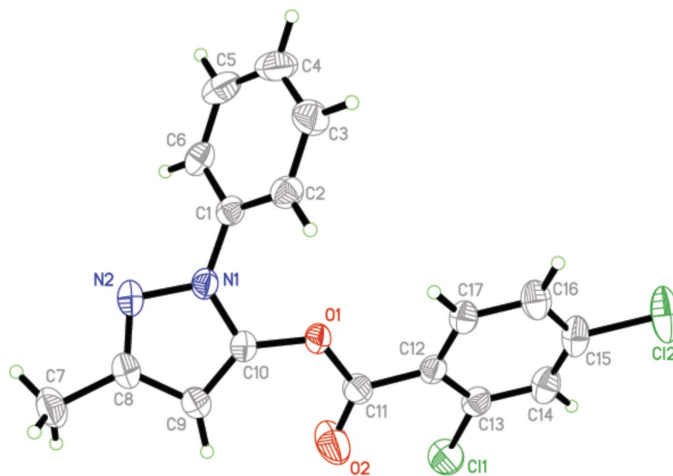


Figure 1

Displacement ellipsoid plot of the molecular structure of (I). H atoms are shown as small spheres of arbitrary radii.

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